# High Speed Reaction Field of Stable/Metastable Bulks Mixed in Nano-Scale and Its Applications

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### [Outline of survey]

Ce-complex oxides with dispersed Pt are well known as an efficient reaction field for automotive exhaust gases. Oxygen transport between stable  $Ce^{3+}_2Zr^{4+}_2O_7$ (pyrochlore phase) and oxygen-intercalated metastable  $Ce^{4+}_2Zr^{4+}_2O_8(\kappa \text{ phase})$  contributes to oxidation of HC and CO and reduction of NOx gas. The metastable  $\kappa$  phase decomposes around 1100°C. For Pr-oxides, there exists several CaF<sub>2</sub>-like, stable phases, which are expressed as  $Pr_nO_{2n-2} = (Pr^{3+}_4Pr^{4+}_m)O_{2(4+m)-2}$  (m=0,1,3,5,6,7,8°°. Two oxygen ions are removed from total oxygen sites twice than cations. m=4 corresponds to  $Ce^{3+}_4Zr^{4+}_4O_{14}=$  $Ce^{3+}_2Zr^{4+}_2O_7$  pyrochlore. Assembly of rare-earth ion with 3+ and metal ions with 4+ may lead to various several CaF<sub>2</sub>-like stable phases and also oxygen-intercalated metastable-phases. In this project, various resembled CaF<sub>2</sub>-like stable and metastable phases are mixed in nano-scale to control the interface energy. The stable/metastable phases should become virtually stable at higher than 1300°C, and make possible high-speed oxygen transfer even at low temperature as 300°C.

# [Expected results]

Several noble complex oxides containing rare-earth elements may be detected. Dream of efficient reaction field, working at low temperature as  $300^{\circ}$ C, and being stable even at  $1300^{\circ}$ C, without addition of Pt, may come true. To control stable/ metastable nano-space leads to various speed and quantity of oxidation-reduction reaction. This property can be applied to hydrogen production via  $C_nH_{2n+2}$  and  $H_2O$ , baiomass and  $H_2O$ , and also selective oxidation of CO in  $H_2$ +CO gases.

### [References by the principal investigator]

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[Term of project] FY2007-2011

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